Q1W4- Test 1- Chemistry of life

Multiple Choice Identify the choice that best co

 1.	Electrons move about the nucleus of an atom:	in reg	gions called			
	a. electron clouds	c.				
	b. nuclei	d.	isotopes			
 2.	When molecules of glucose and fructose combine to form sucrose, they do so by					
	a. hydrolysis	c.	condensation			
	b. electron clouds	d.	radiation			
3.	 The early atmosphere lacked an ozone layer to protect early life forms from the sun's harmful UV rays. Which property of water would allow life to continue under these unfavorable conditions? a. Water is polar, allowing it to dissolve other polar substances. b. Water contains the oxygen necessary for life. c. Water expands when it freezes, which allows a protective layer of ice to form at the surface of a body of water. d. All of the above 					
4.	Which of the following does NOT describe a	polvr	ner?			
	a. Polymers are made of monomers.					
	b. Polymers are large molecules.					
	c. Polymers usually form by covalent bondir	ıg.				
	d. Polymers are broken down by the process	of h	ydrogenation.			
 5.	A very strong base might have a pH of					
	a. 3	c.	9			
	b. 5	d.	13			
 6.	A particle is found to have 8 protons, 10 electrons	and 10 neutrons. What is it?				
	a. Oxygen-20		Oxygen-18			
	b. Oxygen-20 ion with a -2 charge		Oxygen-18 ion with a -2 charge			
 7. Water dissolves many ionic and molecular compounds because of its						
	a. ionic bonding		covalent bonding			
	b. polarity		hydrogen bonding			
 8.	Glucose and fructose, with the formula C_6H_{12}					
	a. numbers of atoms		kinds of atoms			
	b. arrangement of atoms		arrangement of electrons			
 9.	Carbon compounds that come from living organisms are called compounds.					
	a. water		homogeneous			
	b. organic		biological			
 10.	Which of the following can NOT occur during					
	a. New atoms can be made.		New molecules can be made.			
	b. Bonds between atoms can be broken.		Bonds between atoms can be formed.			
 11.	Unlike carbohydrates and fats, proteins contain		` .			
	a. nitrogen		hydrogen			
	b. carbon	d.	oxygen			
 12.	The various enzymes in our bodies are		1 21			
	a. lipids	C.	nucleotides			
	b. carbohydrates	d.	proteins			
 13.	Which of the following is NOT true of trace e	leme	nts?			

	a. They are obtained from foods.b. They are found in very small quantities.c. They are nonessential to most organisms.					
	d. None of the above					
 14.	Carbon-12, carbon-13, and carbon-14 are	_•				
	a. isotopes	c.	radioisotopes			
	b. polymers	d.	macromolecules			
 15.	The nucleus of an atom contains					
	a. protons and neutrons	C.	protons and electrons			
1.0	b. neutrons and electrons	d.	protons, neutrons, and electrons			
 16.	How many electrons can a carbon atom share?	0	three			
	a. one b. two	c. d.	three four			
17.						
 1/.	Which of the images in Figure 6-4 depicts dynamic equilibrium?					
	A B C D					
	Figure 6-4					
	a. A	c.	С			
	b. B	d.	D			
 18.	represents a formula for a chemical com-	poui	nd.			
	a. H		P			
	b. C	d.	H_2O			
 19.	Which of the following is NOT true of lipids?					
	a. They are commonly called fats and oils.					
	b. They are hydrophobic.c. They are good for energy storage.					
	d. They are polar.					
20.	An unsaturated lipid contains					
 _0.	a. more oxygen than hydrogen	c.	ionic bonds			
	b. double bonds	d.	only one fatty acid			
 21.	How does temperature affect the rate of diffusion?					
	a. An increase in temperature causes faster dif	fusi	on by increasing the speed of molecular			
	movement. b. An increase in temperature causes faster difference of the control o	fuci	on by decreasing the speed of molecular			
	movement.	itusi	on by decreasing the speed of molecular			
	c. An increase in temperature causes slower diffusion by eliminating Brownian movement.					
	d. An increase in temperature has no effect on the rate of diffusion.					
 22.	What are the basic building blocks of proteins?					
	a. nucleic acids	c.	amino acids			
	b. peptide bonds		glycerol and fatty acids			
 23.	Which element would need to be removed from	the	molecule in Figure 6-5 to make it unsaturated?			

O Saturated fatty acid group $\begin{array}{c} \text{CH}_2-\text{O-C-CH}_2-\text{CH}$

Figure 6-5

a. carbonb. hydrogenc. oxygend. phosphorus

Matching

Match each item with the correct statement below.

a. cellulose
b. polar molecule
c. nucleus
d. peptide bond
e. polymer
f. solution
g. enzyme
h. metabolism

24. all the chemical changes that occur within an organism

25. center of an atom

26. large molecule formed when many smaller molecules bond together

27. bond formed between amino acids

28. glucose polymer that forms the cell walls of plants

29. protein that speeds up a chemical reaction

____ 30. molecule with unequal distribution of charge